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Citation: [Applied Physics Letters](#) 85, 5860 (2004); doi: 10.1063/1.1830074

View online: <http://dx.doi.org/10.1063/1.1830074>

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Why can CuInSe_2 be readily equilibrium-doped n -type but the wider-gap CuGaSe_2 cannot?

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(Received 24 August 2004; accepted 20 October 2004)

The wider-gap members of a semiconductor series such as diamond \rightarrow Si \rightarrow Ge or AlN \rightarrow GaN \rightarrow InN often cannot be doped n -type at equilibrium. We study theoretically if this is the case in the chalcopyrite family $\text{CuGaSe}_2 \rightarrow \text{CuInSe}_2$, finding that: (i) Bulk CuInSe_2 (CIS, $E_g = 1.04$ eV) can be doped at equilibrium n -type either by Cd or Cl, but bulk CuGaSe_2 (CGS, $E_g = 1.68$ eV) cannot;

Eq. (1), Cu-poor (i.e., low $\Delta\mu_{\text{Cu}}$) and dopant-rich (i.e., maximal $\Delta\mu_{\text{Cd}}$)

tial range of Fig. 1. The “point N ” conditions are defined by $\Delta\mu_{\text{Cu}}=0$, $\Delta\mu_{\text{In}}=-0.07$ eV, and $\Delta\mu_{\text{Se}}=-0.83$ eV in CIS, and by $\Delta\mu_{\text{Cu}}=0$, $\Delta\mu_{\text{Ga}}=-0.21$ eV, and $\Delta > \mu$